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carlyberns. chemistry final, 2012, study guide. Terms in this set (79) Chemical Reaction. process by which atoms of one or more substances are rearranged to form a different substance (indicated by temperature, change in color, odor, gas bubbles, or precipitate) reactants. starting substances.

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Chemistry Final Study Guide. STUDY. PLAY. pure substances. have fixed composition and unique properties. mixtures. composed of 2 or more substances in such a way that each substance retains its chemical identity. types of pure substances. 1. element 2. compound. element.

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Chemistry Final Study guide. STUDY. PLAY. oxidation number. the positive or negative charge of a monatomic ion. atom. the smallest particle of an element that retains all the properties of that element; is electrically neutral, spherically shaped, and composed of electrons, protons, and neutrons.

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